

1 15A NCAC 11 .0303 is amended with changes as published in NCR 27:22, pp. 2031-2073, as follows:

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3 **15A NCAC 11 .0303 EXEMPT CONCENTRATIONS: OTHER THAN SOURCE MATERIAL**

4 (a) No person shall introduce radioactive material into a product or material knowing or having reason to believe
5 that it will be transferred to persons exempt under Paragraph ~~(b)~~ (d) of this Rule or equivalent regulations of the U.S.
6 Nuclear Regulatory Commission or any agreement state, except in accordance with a specific license issued by the
7 Nuclear Regulatory Commission pursuant to ~~Rule .0325 of this Section.~~ 10 CFR 32.11.

8 (b) A manufacturer, processor, or producer of a product or material is exempt from the requirements for a license
9 set forth in ~~{these rules}~~ the rules of this Section to the extent that this person transfers radioactive material
10 contained in a product or material in concentrations not in excess of those specified in ~~{paragraph}~~ Paragraph (d) of
11 this ~~{rule,}Rule, and introduced into the product or material by a licensee holding a specific license issued by the~~
12 US Nuclear Regulatory Commission expressly authorizing such introduction. This exemption does not apply to the
13 transfer of byproduct material contained in any food, beverage, cosmetic, drug, or other commodity designed for
14 ingestion or inhalation by, or application to, a human being.

15 (c) This ~~{rule}Rule shall not be deemed to authorize the import of radioactive material or products containing~~
16 radioactive material.

17 ~~(b)~~ (d) Except as provided in Paragraph (a) and (b) of this Rule, any person is exempt from these Rules to the extent
18 that such person receives, possesses, uses, transfers, owns, or acquires products or materials containing radioactive
19 material in concentrations not in excess of those listed in the following table:

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21 EXEMPT CONCENTRATIONS

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Element (atomic number)	Isotope	Column I	Column II
		Gas concentration microcurie/ml	Liquid and solid concentration microcurie/ml
Antimony (51)	Sb 122		3X10 ⁻⁴
	Sb 124		2X10 ⁻⁴
	Sb 125		1X10 ⁻³
Argon (18)	Ar 37	1X10 ⁻³	
	Ar 41	4X10 ⁻⁷	
Arsenic (33)	As 73		5X10 ⁻³
	As 74		5X10 ⁻⁴
	As 76		2X10 ⁻⁴
	As 77		8X10 ⁻⁴
Barium (56)	Ba 131		2X10 ⁻³

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1		Ba 140		3×10^4
2	Beryllium (4)	Be 7		2×10^2
3	Bismuth (83)	Bi 206		4×10^4
4	Bromine (35)	Br 82	4×10^{-7}	3×10^{-3}
5	Cadmium (48)	Cd 109		2×10^{-3}
6		Cd 115m		3×10^4
7		Cd 115		3×10^4
8	Calcium (20)	Ca 45		9×10^{-5}
9		Ca 47		5×10^4
10	Carbon (6)	C 14	1×10^6	8×10^{-3}
11	Cerium (58) (58)	Ce 141		9×10^4
12		Ce 143		4×10^4
13		Ce 144		1×10^4
14	Cesium (55)	Cs 131		2×10^{-2}
15		Cs 134m		6×10^{-2}
16		Cs 134		9×10^{-5}
17	Chlorine (17)	Cl 38	9×10^{-7}	4×10^{-3}
18	Chromium (24)	Cr 51		2×10^{-2}
19	Cobalt (27)	Co 57		5×10^{-3}
20		Co 58		1×10^{-3}
21		Co 60		5×10^4
22	Copper (29)	Cu 64		3×10^{-3}
23	Dysprosium (66)	Dy 165		4×10^{-3}
24		Dy 166		4×10^4
25	Erbium (68)	Er 169		9×10^4
26		Er 171		1×10^{-3}
27	Europium (63)	Eu 152		6×10^4
28		(T_{1/2} = 9.2 Hrs.) (Half-life = 9.2 Hrs.)		
29		Eu 155		2×10^{-3}
30	Fluorine (9)	F 18	2×10^6	8×10^{-3}
31	Gadolinium (64)	Gd 153		2×10^{-3}
32		Gd 159		8×10^4
33	Gallium (31)	Ga 72		4×10^4
34	Germanium (32)	Ge 71		2×10^{-2}
35	Gold (79)	Au 196		2×10^{-3}
36		Au 198		5×10^4

1		Au 199		2×10^{-3}
2	Hafnium (72)	Hf 181		7×10^{-4}
3	Hydrogen (1)	H 3	5×10^{-6}	3×10^{-2}
4	Indium (49)	In 113m		1×10^{-2}
5		In 114m		2×10^{-4}
6	Iodine (53)	I 126	3×10^{-9}	2×10^{-5}
7		I 131	3×10^{-9}	2×10^{-5}
8		I 132	8×10^{-8}	6×10^{-4}
9		I 133	1×10^{-8}	7×10^{-5}
10		I 134	2×10^{-7}	1×10^{-3}
11	Iridium (77)	Ir 190		2×10^{-3}
12		Ir 192		4×10^{-4}
13		Ir 194		3×10^{-4}
14	Iron (26)	Fe 55		8×10^{-3}
15		Fe 59		6×10^{-4}
16	Krypton (36)	Kr 85m	<u>1×10^{-6}</u>	1×10^{-6}
17		Kr 85	<u>3×10^{-6}</u>	3×10^{-6}
18	Lanthanum (57)	La 140		2×10^{-4}
19	Lead (82)	Pb 203		4×10^{-3}
20	Lutetium (71)	Lu 177		1×10^{-3}
21	Manganese (25)	Mn 52		3×10^{-4}
22		Mn 54		1×10^{-3}
23		Mn 56		1×10^{-3}
24	Mercury (80)	Hg 197m		2×10^{-3}
25		Hg 197		3×10^{-3}
26		Hg 203		2×10^{-4}
27	Molybdenum (42)	Mo 99		2×10^{-3}
28	Neodymium (60)	Nd 147		6×10^{-3} <u>6×10^{-4}</u>
29		Nd 149		3×10^{-4} <u>3×10^{-3}</u>
30	Nickel (28)	Ni 65		1×10^{-3}
31	Niobium (Columbium)(41)	Nb 95		1×10^{-3}
32		Nb 97		9×10^{-3}
33	Osmium (76)	Os 185		7×10^{-4}
34		Os 191m		3×10^{-2}
35		Os 191		2×10^{-3}
36		Os 193		6×10^{-4}

1	Palladium (46)	Pd 103	3×10^{-3}
2		Pd 109	9×10^{-4}
3	Phosphorus (15)	P 32	2×10^{-4}
4	Platinum (78)	Pt 191	1×10^{-3}
5		Pt 193m	1×10^{-2}
6		Pt 197m	1×10^{-2}
7		Pt 197	1×10^{-3}
8	Polonium (84)	Po 210	7×10^{-6}
9	Potassium (19)	K 42	3×10^{-3}
10	Praseodymium (59)	Pr 142	3×10^{-4}
11		Pr 143	5×10^{-4}
12	Promethium (61)	Pm 147	2×10^{-3}
13		Pm 149	4×10^{-4}
14	Radium (88)	Ra 226	1×10^{-7}
15		Ra 228	3×10^{-7}
16	Rhenium (75)	Re 183	6×10^{-3}
17		Re 186	9×10^{-4}
18		Re 188	6×10^{-4}
19	Rhodium (45)	Rh 103m	1×10^{-1}
20		Rh 105	1×10^{-3}
21	Rubidium (37)	Rb 86	7×10^{-4}
22	Ruthenium (44)	Ru 97	4×10^{-3} <u>4×10^{-4}</u>
23		Ru 103	8×10^{-4}
24		Ru 105	1×10^{-3}
25		Ru 106	1×10^{-4}
26	Samarium (62)	Sm 153	8×10^{-4}
27	Scandium (21)	Sc 46	4×10^{-4}
28		Sc 47	9×10^{-4}
29		Sc 48	3×10^{-4}
30	Selenium (34)	Se 75	3×10^{-3}
31	Silicon (14)	Si 31	9×10^{-3}
32	Silver (47)	Ag 105	1×10^{-3}
33		Ag 110m	3×10^{-4}
34		Ag 111	4×10^{-4}
35	Sodium (11)	Na 24	2×10^{-3}
36	Strontium (38)	Sr 85	1×10^{-3} <u>1×10^{-4}</u>

1		Sr 89		1×10^{-4}
2		Sr 91		7×10^{-4}
3		Sr 92		7×10^{-4}
4	Sulfur (16)	S 35	9×10^{-8}	6×10^{-4}
5	Tantalum (73)	Ta 182		4×10^{-4}
6	Technetium (43)	Tc 96m		1×10^{-1}
7		Tc 96		1×10^{-3}
8	Tellurium (52)	Te 125m		2×10^{-3}
9		Te 127m		6×10^{-4}
10		Te 127		3×10^{-3}
11		Te 129m		3×10^{-4}
12		Te 131m		6×10^{-4}
13		Te 132		3×10^{-4}
14	Terbium (65)	Tb 160		4×10^{-4}
15	Thallium (81)	Tl 200		4×10^{-3}
16		Tl 201		3×10^{-3}
17		Tl 202		1×10^{-3}
18		Tl 204		1×10^{-3}
19	Thulium (69)	Tm 170		5×10^{-4}
20		Tm 171		5×10^{-3}
21	Tin (50)	Sn 113		9×10^{-4}
22		Sn 125		2×10^{-4}
23	Tungsten (Wolfram) (74)	W 181		4×10^{-3}
24		W 187		7×10^{-4}
25	Vanadium (23)	V 48		3×10^{-4}
26	Xenon (54)	Xe 131m	<u>4×10^{-6}</u>	<u>4×10^{-6}</u>
27		Xe 133	<u>3×10^{-6}</u>	<u>3×10^{-6}</u>
28		Xe 135	<u>1×10^{-6}</u>	<u>1×10^{-6}</u>
29	Ytterbium (70)	Yb 175		1×10^{-3}
30	Yttrium (39)	Y 90		2×10^{-4}
31		Y 91m		3×10^{-2}
32		Y 91		3×10^{-4}
33		Y 92		6×10^{-4}
34		Y 93		3×10^{-4}
35	Zinc (30)	Zn 65		1×10^{-3}
36		Zn 69m		7×10^{-4}

1		Zn 69	2X10 ⁻²
2	Zirconium (40)	Zr 95	6X10 ⁻⁴
3		Zr 97	2X10 ⁻⁴
4	Beta and/or <u>or</u> gamma emitting		1X10 ⁻¹⁰
5	radioactive material not		1X10 ⁻⁶
6	listed above with half-life		
7	less than 3 years		

9 ~~(e)~~ (e) In Column I of the table, in Paragraph ~~(b)~~ (d) of this Rule, values are given only for those materials normally
10 used as gases.

11 ~~(f)~~ (f) In Column II of the table, in Paragraph ~~(b)~~ (d) of this Rule, the units, microcuries per gram, are used for
12 solids.

13 ~~(g)~~ (g) Many radioisotopes disintegrate into isotopes which are also radioactive. In expressing the concentrations in
14 Paragraph ~~(b)~~ (d) of this Rule, the activity stated is that of the parent isotope and takes into account the daughters.

15 ~~(h)~~ (h) For purposes of this Rule, where a combination of isotopes is involved, the limit for the combination shall be
16 derived as follows: Determine for each isotope in the product the ratio between the concentration present in the
17 product and the exempt concentration established in Paragraph ~~(b)~~ (d) of this Rule for the specific isotope when not
18 in combination. The sum of the ratios shall not exceed unity. An example of this is:

$$\frac{\text{Concentration of Isotope A in Product}}{\text{Exempt concentration of Isotope A}} + \frac{\text{Concentration of Isotope B in Product}}{\text{Exempt concentration of Isotope B}} \text{ less than or equal to } 1$$

27 *History Note:* Authority G.S. 104E-7; 104E-10; 104E-20; 10 CFR 30.70;
28 Eff. February 1, 1980;
29 Amended Eff. October 1, 2013; May 1, 1993; June 1, 1989.